which one as south pole?

(iii) In what ways an electromagnet is better than a permanent magnet?

## Section - B

- 19. What impact will it have if all the organisms in one trophic level are killed?
- 20. Name an organism in which asexual reproduction takes place through budding.
- 21. Name the plant hormones responsible for the following:
  - (a) Elongation of cells.
  - (b) Growth of stem.
- 22. Why are traits acquired during the life time of individual not inherited?
- 23. Mention one function for each of these hormones:
  - (a) Thyroxin
  - (b) Insulin
  - (c) Adrenaline
  - (d) Growth hormone
- 24. What is the advantage of having four chambered heart?
- 25. Why do we insist upon "sustainable natural resource management"? Give any three reasons.
- 26. What are the adaptations of leaf for photosynthesis?
- 27. (i) What are the differences between aerobic and anaerobic respiration? Name some organisms that use the anaerobic mode of respiration.
  - (ii) Describe the structure and functioning of nephrons.



- (i) How are water and minerals transported in plants?
- (ii) Explain the process of nutrition in Amoeba.





Er Manish Bhadoria's Strong Foundation for a bright future

Er Manish Bhadoria's Interactions Study Circle

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## Science

(Sample Paper - II)

Time: 2½ hours Max Marks: 60

## **Marking Scheme**

Q. Nos.	1-6, 19-21	7-12, 22-24	13-16, 25, 26	17-18, 27
Marks	1	2	3	5

## Section - A

- 1. Name a mirror that can give an erect and enlarged image of an object.
- 2. Write the chemical equation to represent the reaction taking place between sodium metal and cold water.
- 3. Why is the difference in colours of the Sun observed during sunrise/sunset and noon?
- 4. Why should we add acid into water for dilution and not water into acid?
- Why is tungsten the preferred element for filament of electric lamps?
- 6. Write the name of the compound in adjoining figure.

- 7. State the factors on which the resistance of a metallic wire depends. Explain how the resistance depends on the factors stated by you.
- 8. Draw the pattern of magnetic field lines around a bar magnet. No two magnetic force lines ever intersect each other. Why?
- 9. A solution of a substance 'X' is used for white washing.
  - (a) Name the substance 'X' and write its formula.
  - (b) Write the reaction of the substance 'X' named in (i) above

with water

- 10. Differentiate between roasting and calcination processes used in metallurgy. Give an example of each.
- 11. What is biogas? Which is the main constituent of biogas? List it's any four characteristics on account of which it is considered an ideal fuel.
- 12. Why is there a need for harnessing non-conventional sources of energy? How can energy be harnessed from the sea in different ways?
- 13. (i) Why detergents are better cleansing agents than soaps? Explain. (ii) What is saponification? Write the reaction involved in this process.
- 14. Explain the structure and functioning of Human eye. How are we able to see nearby as well as distant objects?
- 15. Salt A commonly used in bakery products on heating gets converted into another salt B which itself is used for removal of hardness of water and a gas C is evolved. The gas C, when passed through lime water, turns it milky. Identify A, B and C. Also write the chemical equations involved in the above processes.
- 16. A convex lens of focal length 40 cm is placed in contact with a concave lens of focal length 25 cm. What is the focal length and nature of this combination?
- 17. An element is placed in 2nd Group and 3rd Period of the Periodic Table, burns in presence of oxygen to form a basic oxide.
  - (a) Identify the element.
  - (b) Write the electronic configuration.
  - (c) Write the balanced equation when it burns in the presence of air.
  - (*d*) Write a balanced equation when this oxide is dissolved in water.
  - (e) Draw the electron dot structure for the formation of this oxide.

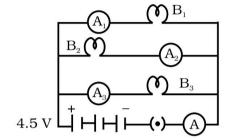


Atomic numbers of a few elements are given below:

10, 20, 7, 14

(a) Identify the elements.

- (b) Identify the Group number of these elements in the Periodic Table. (c) Identify the Periods of these elements in the Periodic Table.
- (d) What would be the electronic configuration for each of these elements?
- (e) Determine the valencies of these elements.
- 18. (i) B<sub>1</sub>, B<sub>2</sub> and B<sub>3</sub> are three identical bulbs connected as shown in Figure. When all the three bulbs glow, a current of 3A is recorded by the ammeter A.



- (a) What happens to the glow of the other two bulbs when the bulb B<sub>1</sub> gets fused?
- (b) What are the reading of A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub> and A when the bulb B<sub>2</sub> gets fused?
- (c) How much power is dissipated in the circuit when all the three bulbs glow together?
- (ii) An electric refrigerator rated 400 W operates 8 hour/day. What is the cost of the energy to operate it for 30 days at Rs 3.00 per kWh?



- (i) Draw the pattern of lines of force due to a magnetic field through and around a current carrying loop of wire. How would the strength of the magnetic field produced at the centre of the circular loop be affected if:
  - (a) the strength of the current passing through the loop is doubled?
  - (b) the radius of the loop is reduced to half of the original radius?
- (ii) When a current is passed through a solenoid, it becomes a magnet. Which end of the solenoid behaves like its north pole and